Limnological characteristics, community metabolism and management strategies of a coastal sinkhole in Cuba (Cenote Jennifer)

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Abstract – The Cenote Jennifer is an important and unique aquatic sinkhole in Cayo Coco (Jardines del Rey Tourist Destination) that has brackish to saline water. Two samplings were made in 1998 and 2009, and 4 metabolism community experiments in 2009. Some limnological parameters were measured in both samplings (temperature, salinity, pH, dissolved oxygen major ions, hydrogen sulfide, nutrients and others). Community metabolism was measured through incubated oxygen concentration in clear and dark oxygen bottles. Results showed that the sinkhole limnology depends on rainfall and light incidence year, with some stratification episodes, due to halocline or oxycline presence, rather than thermocline. The sinkhole water was oligotrophic (total nitrogen of 41.5 ± 22.2 μmol l⁻¹ and total phosphorus of 0.3 ± 0.2 μmol l⁻¹) and with low productivity (gross primary productivity of 63.0 mg C m⁻² d⁻¹). Anoxia and hypoxia were present at the bottom with higher levels of hydrogen sulfide, lower pH and restricted influence of the adjacent sea (2 km away). To protect the Cenote Jennifer, tourist exploitation should be avoided and more resources to ecological and morphological studies should be allocated, and eventually use this aquatic system only for specialized diving. For conservation purposes, illegal garbage disposal in the surrounding forest should end.

Keywords: Limnological / metabolism / management / sinkhole / Cuba

1 Introduction

Approximately 15 percent of the Earth’s land surface is karst. The distribution of karst is essentially the same as the distribution of carbonate rocks, which means that karst terrain occurs mostly in the great sedimentary basins of the world. Karst occurs in North America and the Caribbean region, and is well represented in the Greater Caribbean islands like Hispaniola, Jamaica, Puerto Rico and The Bahamas. It is also well represented in the Yucatan Peninsula (Mexico), in Florida and in the Mississippi basins (USA). In Cuba, karst represents 66% of its total area, including most of the smaller islands and keys of the archipelago (Gutiérrez-Domech, 1998). One of the karstic mesoforms are anchialine caves, partially or totally submerged, located within a few kilometers inland in volcanic or karstic limestone terrain, named sinkholes. Such sinkholes are locally termed “cenotes” in the Yucatan Peninsula in Mexico, “blue holes” in The Bahamas and Belize, and “grietas” in the Galapagos Islands (Iliffe and Kornicker, 2009). In some parts of Cuba, these caves are known as “cenotes”, “dolinas” or “lagunas”.

Around the world, sinkholes have been well studied, particularly in the Yucatan Peninsula (México) (Suárez Morales and Rivera Arriaga, 1998; Schmitter-Soto et al., 2002; Beddows et al., 2007; Sánchez y Pinto et al., 2015; Cervantes-Martínez et al., 2018; Enseñat-Soberanis et al., 2019), Australia (Humphreys, 1999, Humphreys et al., 1999;
Somaratne, 2017), USA (Brinkmann et al., 2008; Emmert 2016; Young et al., 2018) and The Bahamas (Gonzalez et al., 2011; Keeton, 2017; Tamalavage et al., 2018). Brankovits et al. (2017) studied the role of coastal karts subterranean estuaries as sink of methane and its contribution of nutrients and carbon to the ocean.

In Cuba, sinkholes have been poorly studied and the focus has been on biodiversity (Holthuis, 1974; Silva Taboada, 1974; García Debrás et al., 1997; García-Machado et al., 2011; Reynaldo et al., 2016; Pérez-Garcia et al., 2011), morphology (Guarch Rodríguez and Corella Varona, 2010), saline intrusion (Salomon, 2019) and fecal contamination (Mulec and Oarga, 2014).

The geology of Cayo Coco consists of carbonate rocks of the upper-middle Pleistocene (calcarenites and biocalcarenites), belonging to the Jaimanitas formation (Iturralde-Vinent, 1994). There are many karstic forms, but only one sinkhole is known (Cenote Jennifer).

There are only two references to the Cenote Jennifer (G’meiner, 2016; Peros et al., 2017), both regarding paleolimnological results from a sediment core, but they were the first to show some limnological parameters (salinity, temperature, dissolved oxygen and pH); the sinkhole area and bathymetry were also described by these authors. This sinkhole was also studied in 1998 and 2009, mainly to determine limnological parameters. Cayo Coco is located at the Jardines del Rey tourist destination, where the creation of new products like nature tourism, ecotourism and health tourism, is essential for economic success.

The objective of this work is to analyze some limnological parameters and primary productivity of the Cenote Jennifer, and based on such results; provide conservation and management considerations for tourist and scientific use.

2 Materials and methods

2.1 Study area

The geology of Cayo Coco consists of carbonate rocks of the upper-middle Pleistocene (calcarenites and biocalcarenites), belonging to the Jaimanitas formation (Iturralde-Vinent, 1994). Cayo Coco has been tectonically stable throughout the Holocene (Iturralde-Vinent, 1994). Most of the soils of Cayo Coco are shallow and poorly developed on a limestone substrate (Alcolado et al., 2007). The study site (Cenote Jennifer), is located in exposed limestone on the northeast of Cayo Coco, at approximately 2 Km from the coast (Fig. 1).

In Cayo Coco, dry conditions prevail from December to April (<60.0 mm rain per month), and wetter conditions (>100.0 mm rain per month) from May to November, with a marked mid-summer drought (MSD) in July-August. Monthly average temperature ranges from 23.0 °C in January to 29.0 °C in July (Batista Tamayo et al., 2006).

Historically, Cayo Coco has been virtually devoid of human population. Social and economic intervention began at the beginning of the 20th century. It was characterized mainly by forest exploitation and charcoal production; in the adjacent coast, fishing was the principal activity. Cattle ranching was also tried for some time, which is why feral cattle still inhabit Cayo Coco. Many forest areas were cleared for timber harvesting or burned for pasture lands (Alcolado et al., 1998). In most cleared areas, secondary forests have developed. In the early 1980s several studies to assess the potential use of the area for tourism were conducted. In 1980, the construction of roads on the island began and in the summer of 1986, the construction of a causeway started. It was completed by 1988. For the first time, Cayo Coco was connected to the mainland.
and it made possible the development of tourism infrastructure. In 1990, the first master plan for the development of the area was implemented. By 1993, a large tourist complex had been built on the northeast shore of Cayo Coco (1.5 km from the Cenote Jennifer) and a nearby second hotel was completed and opened on December 20, 1996. By 2016, almost 4000 rooms for tourism only in Cayo Coco and more than 5000 rooms in the province of Ciego de Ávila had been built (ONEI, 2018).

Two decades before, tourism authorities in search for nature tourism options, had designed a trail named “Las Dolinas”, where the Cenote Jennifer was included as the main attraction, but it never yielded the expected results.

The Cenote Jennifer is located (22°31’50” N, 78°22’58” W) in Cayo Coco, province of Ciego de Ávila, Cuba (Fig. 1). It has a surface area of approximately 400 m² (0.04 ha) and a depth of 17 m from the highest point of the surrounding rock surface to the sediment-water interface. Depth ranges from 9 m near the edge of the sinkhole to 15 m at the center (G’meiner, 2016).

The Cenote Jennifer can be classified as a typical sinkhole in the form of a glass (variant form with attached flooded galleries), according to classification given by Hall (1936) and as a sinkhole in a coastal line, according to the criteria of Navarro-Mendoza (1988).

The dense forest directly surrounding the sinkhole is characterized as a thorny limestone shrub wood, consisting of Picrodendron baccatum (Jamaican walnut), and Bursera simaruba (gumbo-limbo) trees and a shrub layer of Buxus spp., Randia spp. and Croton spp. Approximately 50 m to the north of the Cenote Jennifer is a shallow hypersaline lagoon (Pupi II), fringed by Avicennia germinans and Conocarpus erectus (G’meiner, 2016).

The direction of the groundwater flow in Cayo Coco is quite variable, usually radial, with from points of greater dimensions towards its central zone. The absolute levels of water vary from 0.11 to 0.57 meters in the wet period and from 0.04 to 0.44 m during the dry season, except in areas of possible tectonic dislocation, where static underground water levels are below average sea level.

The wastewater treatment system (stabilization lagoons) of Cayo Coco is near the sinkhole (800 m) (Fig. 1). It has effluent infiltration to the ground as tertiary and final treatment. This final effluent disposal dates from the beginning of the present century and could affect groundwater in Cayo Coco.

2.2 Sampling and analysis

Two separate samplings were made at the sinkhole. The first sampling took place in January and February of 1998. Samples (4) were taken from the surface and at depths of 4 m, 8 m and 12 m. Temperature, pH, and salinity were measured in situ (with an ORION probe) and samples for dissolved oxygen, nutrients (ammonium), total alkalinity, total hardness, carbon dioxide, hydrogen sulphide, some ions (calcium, magnesium, potassium, chloride and sodium), and soluble reactive silicate (SRSi) were taken to the laboratory. All samples were preserved according to the methodology for each parameter. The methodologies proposed by AHPA (1985) for each parameter were used in the laboratory. Major ions were determined by Atomic Absorption Spectrometry (K⁺: DL = 0.1 mg/L; Ca²⁺: DL = 0.002 mg/L; Mg²⁺: DL = 0.001 mg/L; Na⁺: DL = 0.1 mg/L), total hardness by EDTA titration (DL = 0.1 mg/L), total alkalinity by titration with HCL (0.1 mg/L), hydrogen sulfide by iodometric method (DL = 0.1 mg/L) and chloride by argentometric method (DL = 0.1 mg/L). Dissolved oxygen was determined by Winkler method (DL = 0.1 mg/L). Ammonium was determined with automatic method (phenate method, DL = 0.14 μM/L) R and SRSi were determined automatically through molybdate method (DL = 0.01 μM/L) and total nitrogen and phosphorus was determined by persulfate reduction (DL = 0.42 μM/L and 0.02 μM/L, respectively).

Four samplings were made in March (12 and 19) and in April of 2009 (28 and 30). Surface water temperature and salinity (with digital thermo-salinometer WLW), dissolved inorganic nutrients (ammonium, nitrate + nitrite), total nutrients (nitrogen and phosphorus) and soluble reactive silicate (SRSi) were measured before and after every metabolism experiment. All samples were preserved using the methodology for each parameter. The methodologies proposed by AHPA (1985) for each parameter were used in the laboratory. Dissolved oxygen was determined by Winkler method (DL = 0.1 mg/L). Nitrate + nitrite through automatic method (cadmium reduction technique, DL = 0.02 μM/L) and soluble reactive phosphorus by molybdate method (DL = 0.01 μM/L). Ammonium was determined with automatic method (phenate method, DL = 0.14 μM/L) R and SRSi were determined automatically through molybdate method (DL = 0.01 μM/L) and Total Nitrogen and Phosphorus were determined by persulfate reduction (DL = 0.42 μM/L and 0.02 μM/L, respectively).

Community metabolism was measured by evaluating the evolution of oxygen dynamics, using the light and dark bottles incubation method. The bottles were incubated for four to six light-hours at depths of 0, 1, 3, 5, 7, 9 and 11 m. At each depth, nine oxygen bottles were filled (three for initial oxygen determination, three for light incubation and three for dark incubation). While sampling, extreme precautions were taken to completely avoid bubbling that could alter oxygen content, as recommended by Valdés-Pino-Castillo et al. (2014). Dissolved oxygen concentration in each bottle was determined in the laboratory, in triplicate for each sample bottle to minimize and assess error (see Valdés-Pino-Castillo et al., 2014 for further details on the method). To determine dissolved oxygen concentration (DO), the method of Winkler, modified by Carritt and Carpenter (UNESCO, 1983) was used.

2.3 Metabolism calculation

Gross primary production (GPP), net primary production (NPP), and community respiration (R) were calculated using the oxygen change rate in the light and dark bottles, respectively, following Wetzel and Likens (1991), and therefore thereafter dividing the differences between initial and final oxygen concentrations by the specific incubation time of each set of bottles.

Conversion of oxygen rates to carbon rates was performed with the theoretical and most widely used conversion values PQ = 1.3 and RQ = 1.0 (Gazeau et al., 2005).
2.4 Density of use

To estimate the maximum density of use (MDU), the total public use area of the sinkhole was divided by 4 m², which is the estimated vital area (VA) for a visitor to feel comfortable in recreational spaces (García Hernández, 2001)

$$MDU = \frac{WM + RA}{VA}$$

(1)

where MDU (Maximum density of use); WM (Water mirror); RA (Rest area); VA (Vital area).

3 Results

3.1 Limnological behavior

All mean values of measured parameters, correlation and ANOVA results in the Cenote Jennifer (in 1998 and 2009) are shown in Table 1. PCA analysis (for measured parameters in 1998) shows four distinct layers; surface layer, first layer (4 m), medium layer (8 m) and bottom and anoxic layers (Fig. 2).

3.1.1 Temperature and salinity

Mean temperature was higher at surface ($F = 160.44$, $p < 0.05$) and declined as depth decreased, but it was not clear thermocline and the difference between surface temperature and temperature at 12 m was only of 1.9°C in 1998 (Fig. 3A). Temperature was different for all samplings (in 1998 and 2009), with the highest value at surface in 1998 (29.6°C). Temperature at 12 m was only of 1.9°C in 1998 (Tab. 1). Sulfate showed a more significant decrease than hydrogen sulphide, which dropped sharply from 5337.2 ± 129.1 mg l⁻¹ at 4 m to 4135.3 ± 430.6 mg l⁻¹ at 12 m (Tab. 1).

3.1.2 Hydrogen ion concentration (pH) and dissolved oxygen (DO)

Mean pH was higher ($F = 154.99$, $p < 0.05$) in surface water (8.08 ± 0.01) and declined (significantly) for each sampled depth in 1998 (Tab. 1; Fig. 3C). DO profile showed a clear oxycline, which began between 4 and 8 m, probably at 6 m, because some occasional measurements showed Secchi transparency of 6–8 m. At surface, mean DO was 7.2 ± 0.2 mg l⁻¹ (Tab. 1; Fig. 3D), while at 8 m was 5.7 ± 0.2 mg l⁻¹ and there was anaerobic at 12 m in 1998. All samplings (in 1998 and 2009) were significantly different, with the highest value at surface (10.3 mg l⁻¹) in the sampling of March 2 (Tab. 1). There was hypoxia at 9 m and at 1 m in the second sampling of April (Fig. 4). DO pro

3.1.3 Hydrogen sulphide (H₂S) and sulfate (SO₄²⁻)

Mean hydrogen sulphide was higher ($F = 1205.85$, $p < 0.05$) at 12 m and increased from 150.0 ± 2.0 mg l⁻¹ at surface to 207.9 ± 3.4 mg l⁻¹ at 12 m in 1998 (Tab. 1). Sulfate showed a more significant decrease than hydrogen sulphide, which dropped sharply from 5337.2 ± 129.1 mg l⁻¹ at 4 m to 4135.3 ± 430.6 mg l⁻¹ at 12 m (Tab. 1).

3.1.4 Dissolved inorganic nitrogen (ammonium)

This nitrogen form increased significantly ($F = 130.06$, $p < 0.05$) from 2.9 ± 0.05 μmol l⁻¹ at surface and 4 m to 7.9 ± 0.09 μmol l⁻¹ at 12 m in 1998. Ammonium had significant (positive and negative) correlations with most measured parameters (Tab. 1) in 1998, but this behavior was different in the 2009 samplings. Only the samplings of April 2009 were similar and with the highest value of ammonium at surface (4.5 μmol l⁻¹) was recorded in the first sampling of 2009 (March 1) (Tab. 1).

3.1.5 Soluble reactive silicate (SRSi)

Mean SRSi had the highest concentration at 12 m (231.8 ± 0.3 μmol l⁻¹), increasing from surface 104.8 ± 0.3 μmol l⁻¹ to 231.8 ± 0.3 μmol l⁻¹ in 1998. The highest SRSi concentration at surface was in 1998 and only the samplings of April of 2009 were similar (Tab. 1). These concentrations are higher than those reported for seawater (less than 3.0 μmol l⁻¹) and are normal for groundwater.

3.1.6 Major ions

Major ions (in meq l⁻¹) showed the following order: Cl⁻ > Na⁺ > Mg²⁺ > SO₄²⁻ > Ca²⁺ > K⁺. All major ions had significant correlation among them (Tab. 1). K⁺, Mg²⁺, Na⁺ and Cl⁻ increased significantly from surface to 12 m; while Ca²⁺ and SO₄²⁻ decreased with depth (Tab. 1).

3.1.7 Total hardness and alkalinity

Total hardness increased significantly from surface to 12 m and had a significant correlation with total alkalinity, ammonium and major ions. Total alkalinity showed a pattern similar to that of total hardness, increasing from surface to 12 m, and with significant correlation with ammonium and major ions (Tab 1).

3.1.8 Dissolved inorganic nitrogen (DIN)

All dissolved inorganic nitrogen forms were measured in 2009; ammonium was the principal fraction of DIN (more than 80% for all samplings). Mean nitrate + nitrite concentration was higher ($F = 10.17$, $p < 0.05$) in March (0.7 ± 0.13 μmol l⁻¹) samplings than in April samplings (0.5 ± 0.063 μmol l⁻¹) (Tab. 1). Mean DIN concentration was (4.7 ± 0.53 μmol l⁻¹), with higher ($F = 6.45$, $p < 0.05$) mean concentration in March than in April. DIN showed a significant correlation only with ammonium in 2009 (Tab. 1).

3.1.9 Total nitrogen (TN)

Most nitrogen in the sinkhole was in organic form, DIN only represented between 7 and 21% of TN. Mean TN concentration was 41.5 ± 22.23 μmol l⁻¹ for all the samplings, with greatest concentration in the first sampling of 2009 (Tab. 1).
Table 1. Mean values of each measured parameter per depth in Cenote Jennifer in 1998 and 2009. Minor bold letters denote marked correlation among parameters. Major letters denote significant differences ($p < 0.05$) only in surface samples among samplings.

<table>
<thead>
<tr>
<th>Depth sampling</th>
<th>Salinity</th>
<th>Temperature (°C)</th>
<th>pH</th>
<th>Dissolved Oxygen (mg l$^{-1}$)</th>
<th>Total Hardness (mg L$^{-1}$)</th>
<th>Total Alkalinity (mg l$^{-1}$)</th>
<th>1998</th>
<th>Soluble Reactive Silicate (μmol l$^{-1}$)</th>
<th>Ammonium (μmol l$^{-1}$)</th>
<th>K$^+$ (mg/L)</th>
<th>Hydrogen Sulfide (mg l$^{-1}$)</th>
<th>Ca$^{2+}$ (mg l$^{-1}$)</th>
<th>Mg$^{2+}$ (mg l$^{-1}$)</th>
<th>SO$_4^{2-}$ (mg l$^{-1}$)</th>
<th>Na$^+$ (mg l$^{-1}$)</th>
<th>Cl$^-$ (mg l$^{-1}$)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Surface</td>
<td>7.0$^a$</td>
<td>29.6$^b$</td>
<td>8.08</td>
<td>7.2$^b$</td>
<td>390.8</td>
<td>64.9</td>
<td>104.8$^b$</td>
<td>2.9$^b$</td>
<td>155.4</td>
<td>152.7</td>
<td>114.1</td>
<td>5337.2</td>
<td>3748.2</td>
<td>7439.2</td>
<td></td>
<td></td>
</tr>
<tr>
<td>4.0 m</td>
<td>12.5</td>
<td>29.0</td>
<td>7.33</td>
<td>6.8</td>
<td>390.8</td>
<td>64.9</td>
<td>104.8</td>
<td>2.9</td>
<td>155.4</td>
<td>152.7</td>
<td>114.1</td>
<td>5337.2</td>
<td>3748.2</td>
<td>7439.2</td>
<td></td>
<td></td>
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<tr>
<td>8.0 m</td>
<td>14.0</td>
<td>28.8</td>
<td>7.27</td>
<td>5.7</td>
<td>426.1</td>
<td>71.1</td>
<td>118.0</td>
<td>4.5</td>
<td>155.2</td>
<td>160.5</td>
<td>1409.9</td>
<td>4483.4</td>
<td>4072.2</td>
<td>8367.4</td>
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<tr>
<td>12.0 m</td>
<td>15.0</td>
<td>27.7</td>
<td>7.19</td>
<td>0.0</td>
<td>465.4</td>
<td>77.0</td>
<td>231.8</td>
<td>7.9</td>
<td>174.8</td>
<td>207.9</td>
<td>1743</td>
<td>4135.3</td>
<td>4373.1</td>
<td>8942.8</td>
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<tr>
<td>Marked correlation</td>
<td>$a$</td>
<td>$b$</td>
<td>$a$</td>
<td>$b,c$</td>
<td>$c,d$</td>
<td>$d,e$</td>
<td>$c$</td>
<td>$b,c,d,e,f$</td>
<td>$d,e$</td>
<td>$d,e$</td>
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<td>$d,e$</td>
<td>$d,e$</td>
<td>$d,e$</td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

Depth sampling | Salinity | Temperature (°C) | pH | Dissolved Oxygen (mg l$^{-1}$) | Soluble Reactive Phosphorus (μmol l$^{-1}$) | Total Phosphorus (μmol l$^{-1}$) | 1998 | Soluble Reactive Silicate (μmol l$^{-1}$) | Ammonium (μmol l$^{-1}$) | Nitrate + nitrite (μmol l$^{-1}$) | Dissolved Inorganic Nitrogen (μmol l$^{-1}$) | Total Nitrogen (μmol l$^{-1}$) |
<table>
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<tbody>
<tr>
<td>Surface (March 1)</td>
<td>15.0$^b$</td>
<td>25.3$^c$</td>
<td>–</td>
<td>6.4$^c$</td>
<td>0.1$^b$</td>
<td>0.6$^a$</td>
<td>$4.5^a$</td>
<td>0.7$^a$</td>
<td>5.2$^c$</td>
<td>77.0$^a$</td>
<td>–</td>
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<td>–</td>
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<tr>
<td>Surface (March 2)</td>
<td>16.0$^c$</td>
<td>26.3$^c$</td>
<td>–</td>
<td>10.3$^c$</td>
<td>ND</td>
<td>0.2$^b$</td>
<td>$60.0^c$</td>
<td>0.5$^c$</td>
<td>5.1$^b$</td>
<td>24.5$^b$</td>
<td>–</td>
<td>–</td>
<td>–</td>
</tr>
<tr>
<td>Surface (April 1)</td>
<td>13.1$^b$</td>
<td>27.5$^d$</td>
<td>–</td>
<td>6.2$^b$</td>
<td>0.03$^b$</td>
<td>0.4$^a$</td>
<td>$53.0^b$</td>
<td>0.5$^b$</td>
<td>4.2$^b$</td>
<td>33.0$^b$</td>
<td>–</td>
<td>–</td>
<td>–</td>
</tr>
<tr>
<td>Surface (April 2)</td>
<td>13.8$^d$</td>
<td>27.4$^d$</td>
<td>–</td>
<td>5.5$^d$</td>
<td>0.1$^a$</td>
<td>0.3$^a$</td>
<td>$56.0^b$</td>
<td>0.5$^a$</td>
<td>4.2$^b$</td>
<td>31.7$^d$</td>
<td>–</td>
<td>–</td>
<td>–</td>
</tr>
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<td>Surface (mean)</td>
<td>14.5</td>
<td>26.6</td>
<td>–</td>
<td>7.1</td>
<td>0.07$^c$</td>
<td>0.3$^d$</td>
<td>$58.5^c$</td>
<td>4.1$^b$</td>
<td>4.7$^b$</td>
<td>41.5$^c$</td>
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<td>$b$</td>
<td>$a$</td>
<td>$c$</td>
<td>$d$</td>
<td>$a,e$</td>
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<td>$f$</td>
<td>$d,e$</td>
<td>–</td>
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Fig. 2. PCA analysis for all measured parameters in Cenote Jennifer in 1998.

Fig. 3. Vertical profiles of (A) temperature (°C), (B) salinity, (C) pH and (D) dissolved oxygen (mg/L) in cenote Jennifer in 1998 samplings.
3.1.10 Soluble reactive phosphorus (SRP) and total phosphorus (TP)

Most TP was organic. SRP only represented between 0 and 40% of TP. Mean SRP was $0.07 \pm 0.063 \mu \text{mol l}^{-1}$, with undetectable concentration in the second sampling of March of 2009 (Tab. 1). Mean TP was $0.3 \pm 0.23 \mu \text{mol l}^{-1}$ for all the sampling period (Tab. 1).

### 3.2 Community metabolism

Vertical profiles of gross primary production (GPP), respiration (R) and Net primary production (NPP) in mg O$_2$ m$^{-3}$ h$^{-1}$ are shown in Figure 5. Mean production (both net and gross) rates were different for each sampling; however, there were no significant differences among experiments. In the first experiment of March (0.29 mg O$_2$ m$^{-3}$ h$^{-1}$) and in the second experiment of April (0.21 mg O$_2$ m$^{-3}$ h$^{-1}$), maximum GPP was documented at 5 m, while in the second experiment of March (0.61 mg O$_2$ m$^{-3}$ h$^{-1}$) it was found at 1 m. In the first experiment of April (0.37 mg O$_2$ m$^{-3}$ h$^{-1}$), maximum GPP was found at surface. Maximum mean GPP (for all experiments), was of 0.29 mg O$_2$ m$^{-3}$ h$^{-1}$ at 1 m. NPP behavior was similar to that of GPP for each experiment, except for the second experiment of April (0.14 mg O$_2$ m$^{-3}$ h$^{-1}$) with the maximum at 11 m and maximum mean NPP was found at surface (0.14 mg O$_2$ m$^{-3}$ h$^{-1}$). Respiration also had a vertical gradient, and maximum for most experiments was found at 5 m. Only in the first experiment of April (0.35 mg O$_2$ m$^{-3}$ h$^{-1}$) the maximum was recorded at 1 m.

Mean R was 0.20 mg O$_2$ m$^{-3}$ h$^{-1}$ at 5 m. After vertically integrating the whole production layer, including experiments at all depths, we calculated vertically integrated metabolic rates (Tab. 2). For each experiment month, values of integrated metabolic rates were similar, without significant differences.

In terms of carbon fluxes, GPP had a mean value of 63.0 mg C m$^{-2}$ d$^{-1}$ and mean NPP was 23.0 mg C m$^{-2}$ d$^{-1}$. The potential carbon exportation from the production layer through biomass sinking (f = NPP/GPP) was 37% in the Cenote Jennifer. Mean R for all experiments was 31.0 mg C m$^{-2}$ d$^{-1}$. The ratio PB:R = 2.0 showed that the prevailing processes at the sinkhole are autotrophic rather than completely heterotrophic.

### 4 Discussion

#### 4.1 Limnological parameters and community metabolism

Cenote Jennifer has a circular shape and is an open sinkhole according to the classification proposed by Hall (1936).

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**Table 2.** Integrated metabolic rates (vertically) in Cenote Jennifer in 2009 experiments.

<table>
<thead>
<tr>
<th>Community metabolism</th>
<th>March 1 (mg O$_2$m$^{-3}$ h$^{-1}$)</th>
<th>March 2 (mg O$_2$m$^{-3}$ h$^{-1}$)</th>
<th>April 1 (mg O$_2$m$^{-3}$ h$^{-1}$)</th>
<th>April 2 (mg O$_2$m$^{-3}$ h$^{-1}$)</th>
<th>Mean</th>
</tr>
</thead>
<tbody>
<tr>
<td>GPP</td>
<td>0.002</td>
<td>0.003</td>
<td>0.002</td>
<td>0.001</td>
<td>0.002</td>
</tr>
<tr>
<td>NPP</td>
<td>0.001</td>
<td>0.001</td>
<td>0.000</td>
<td>0.001</td>
<td>0.001</td>
</tr>
<tr>
<td>R</td>
<td>0.000</td>
<td>0.002</td>
<td>0.002</td>
<td>0.001</td>
<td>0.001</td>
</tr>
</tbody>
</table>
It is a small sinkhole (< 1 ha) that may be unique in the Cayo Coco area. Cenote Jennifer could be classified as a lentic sinkhole according to Beddows et al. (2006). There is not national inventory of large, medium or small sinkholes in Cuba, but it is known that these aquatic systems are distributed throughout the country (including the main and smaller islands of the archipelago). Some authors state that the principal areas of sinkhole are located in the karstic zones of Pinar del Río, the Zapata Swamp and south of Isla de la Juventud (Núñez Jiménez et al., 1968; Franco and De la Torre, 1980; Gutiérrez-Domech, 1998), but the presence of sinkhole have been reported in Holguín (Echtinger, 2000; Guarch and Corella, 2010) and in karstic zones of Havana and Central Cuba (Gutiérrez-Domech, 1998).

Vertically, the sinkhole did not show a clear thermocline between surface and 12 m (variation of 1.9°C) in 1998 and showed similar behavior as observed by Peros et al. (2017), variation of 4.0°C between surface and 14 m. The small difference between samplings (1998 and 2014) could be due to the sampling date (February in 1998, dry and cold season) and in 2017, sampling was performed in July, when mean air temperature is higher in Cayo Coco (wet and warm season). Similar behavior was observed at the Cenote Tanque Azul, in the province of Holguín; with a temperature variation of 3.0°C of thermocline between surface and 41 m (Echtinger, 2000). Cervantes-Martínez et al. (2002) found mixed results with thermal stratification, but slightly or no temperature differences in the water column in eight sinkholes of the Yucatan Peninsula. The presence of thermocline was found by Herrera-Silveira and Comin (2000) in lotic and lentic sinkholes during the dry and rainy seasons, while the water column remained mixed during the winter storm season in Yucatán. One factor that could explain the lack of a clear thermocline (even in the warm season) at the Cenote Jennifer is its location. It is surrounded by a dense forest, which limits the direct incidence of the sun light, which results in less transmission of radiant heat between the atmosphere and its waters (Díaz–Arce et al., 2001; Schmitter-Soto et al., 2002). The thermal stratification process in lakes, for example, depends of diverse factor as wind influence, water movements and lake form (Wetzel, 2001), so, Cenote Jennifer is protected (by a dense forest) from winds and direct light incidence during all day and this could be a principal factor that drives its thermal behavior.

Surface salinity was lower (7.0 ± 0.7) in the 1998 sampling than in that of 2009 (14.5 ± 1.2) according to Peros et al. (2017). Intense precipitation events in Cayo Coco could have influenced this behavior in January and February of 1998. The total amount of rainfall in January (170.6 mm) is the second highest for this month according to the data from the Cayo Coco weather station. Total amount of rainfall of February 1998 (114.1 mm) also ranks second, slightly below the 128.1 mm of 1993 (historic record) for the said station. The sinkhole is located 2 km away from the coast, which could prevent stronger saline intrusion. In the Yucatan Peninsula, salinity in most of the sinkholes surveyed was lower than in the Cenote Jennifer, including the ones near the coast (Elias Gutiérrez et al., 2007; Camargo-Guerra et al., 2013). However, some sinkholes like the Casa Cenote (located on the coast near Tulum, Mexico) is characterized by an active exchange with marine water and had evident salinity differences between the upper and lower layer (Sánchez et al., 2002). Echtinger (2000) found salinity differences of 20.0 between surface and bottom with clear halocline after 16 m at a Cuban coastal sinkhole. Only Peros et al. (2017) described a halocline (between 12 and 14 m), with salinity of 25.07 at the bottom (salinity of Cayo Coco marine waters ranges between 36.0 and 37.5). Groundwater in Cayo Coco is characterized by high salinities (between 10.0 and 20.0) at approximately 0.93 m deep (Hernández Valdés, 2011). So, salinity at the Cenote Jennifer depends mainly on rain volumes and on the weak tide influence near the bottom (Kovacs et al., 2017). For the zone in the Yucatan Peninsula that is 0–0.4 km away from the coast, Beddows (2004) documented that due to low hydraulic conductivity (restricted size of the conduits), the salinity gradient is steep (referred to as mixing zone; Beddows, 2004). Whereas, in the zone located >0.4–10 km away from the coast (and in areas of high conduit density), the gradient is less extreme and the halocline position was lower than predicted by the Ghyben-Herzberg principle (Beddows, 2004).

The increase of concentrations of some major ions from surface to bottom corroborated that the marine environment has some influence over the sinkhole, because ions (K⁺, Mg²⁺, Na⁺ and Cl⁻) are more abundant in seawater than in other naturals waters (Fagundo and González, 2005). Dissolved oxygen concentration was very variable in Jennifer, highly depending on precipitation and supply of organic matter from the surrounding forest. It was significant that in the samplings in January and February of 1998 there was hypoxia at 12 m, while in the 2009 samplings, this condition occurred in the second sampling of April. In these three samplings, rain was typical of the dry season. Brankovits et al. (2017) found that rainfall was the key external factor regulating electron acceptor availability in the meteoric portion of one aquifer in the Yucatán Peninsula.

Peros et al. (2017) found lower DO concentration (1.8 mg l⁻¹) than in our samplings, with a clear oxycline after depth of 5 m, but this sampling was in June of 2014 (wet season). During rainfall events, the organic matter that comes to the sinkhole is oxidized by aerobic heterotrophs that use oxygen (remineralization process), a normal behavior for lentic sinkholes (Schmitter-Soto et al., 2002; Cervantes-Martínez et al., 2002; Beddows et al., 2006, Ramos et al., 2017). Most of DO profiles (including of Peros et al., 2017) coincided that higher DO values are between 3 and 5 m, which could be related to higher photosynthetic production due to primary producers, with a high light-related variability. The Cenote Jennifer is surrounded by a dense forest, a light-limiting factor to primary producers (Díaz–Arce et al., 2001; Beddows et al., 2006). The significant correlation (negative) of DO with sulfide and CO₂ (negative) is an expression of the biogeochemical process at the sinkhole. If dissolved oxygen is completely consumed by aerobic heterotrophs, alternate electron acceptors (e.g., sulfate from seawater or nitrate from meteoric groundwater) may be available for microbes to continue degrading organic matter trapped within the mixing zone (Ruberg et al., 2005; Pohlman, 2011; Brankovits et al., 2017).

In sinkholes with little human activity and with predominat supply of organic matter from the surrounding forest, nutrient concentrations were low, mainly because of rainfall volumes (Cervantes-Martínez et al., 2002; Schmitter-Soto et al., 2002; Elias Gutiérrez et al., 2007,
4.2 Management and conservation considerations

Sinkholes were very important to the indigenous peoples of the Caribbean islands and the Yucatan Peninsula, where they were used both, as sources of freshwater and also for ritualistic activities (Keegan and Carlson, 2008; Cooper, 2010). However, in Cayo Coco there are no archeological evidences of Cuba’s aboriginal peoples. The Cenote Jennifer treasures rich paleo and biogeochemical records in its sediments. In this scenario, and taking into account the importance of paleoclimate studies to explain whether anthropogenic cause for Climate Change, we consider that this sinkhole must be preserved. The first step must be to keep it as tourist unexploited site and then, corroborate the existence of anchialine caves connected to the sinkhole. If so, cave diving could be a great option for specialized tourism as another product of the Jardines del Rey destination, particularly Cayo Coco.

Maximum density of use (MDU) calculated for Cenote Jennifer, taking into account 4 m² of estimated vital area and a rest area of 30 m², was of 107 visitors simultaneously. However, this sinkhole is not used as a swimming or bathing site in Cayo Coco, because Cayo Coco has beautiful sandy beaches. Many of the sinkholes in the Yucatan Peninsula, currently used for swimming and other tourist activities are over exploited and in dangerous conditions due to chemical, fecal and garbage contamination (Enseñat-Soberanis et al., 2019).

Illegal disposal of garbage is one of the major environmental problems in Cayo Coco and in the forest surrounding the Cenote Jennifer. This situation must be managed to avoid future contamination of this aquatic resource. Currently, there is no evidences of possible chemical contamination from infiltration through the aquifer of sewage treated in a nearby plant. Undoubtedly, this is good news for the sinkhole conservation strategies.

5 Summary

The Cenote Jennifer is an important and unique aquatic site in Cayo Coco (Jardines del Rey Tourist Destination) that has brackish to euhaline water, depending on the season and with some degree of communications with marine water. It is a lentic and oligotrophic system with a very low productivity and most of the time with anoxic waters at the bottom. However, it is of great scientific importance as evidence of paleoclimate records have been found in its sediments. The sinkhole is subject to some threats like garbage disposal in the surrounding forest and probable over exploitation for tourist purposes. At present, its conservation status is good and must be preserved as a scientific and specialized diving site in the future.

References


Sinkhole on Andros Island, the Bahamas. Doctoral dissertation. Texas A&M University, USA.


